

# Structural and Magnetic Properties of a Novel Double Perovskite $\text{Sm}_2\text{MgRuO}_6$

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**Abstract.** Double perovskite oxides ( $\text{A}_2\text{B}'\text{B}''\text{O}_6$ ) have attracted significant attention due to their unique combination of functional properties. These materials are particularly promising for spintronic and energy applications, where precise control of cation ordering and magnetic interactions is critical. In this study, we synthesized the novel double perovskite  $\text{Sm}_2\text{MgRuO}_6$  via solid-state reaction and investigated its structural and magnetic properties to understand its potential for advanced technologies. The crystal structure of  $\text{Sm}_2\text{MgRuO}_6$  was identified by X-ray diffraction (XRD) at room temperature and Rietveld refinement confirm that the compound crystallizes in the monoclinic  $P2_1/n$  space group with lattice parameters  $a = 5.41194 \text{ \AA}$ ,  $b = 5.66563 \text{ \AA}$ , and  $c = 7.71585 \text{ \AA}$ . The structure exhibits a rock-salt-like cationic ordering of  $\text{Mg}^{2+}$  and  $\text{Ru}^{4+}$  on the octahedral sites, similar to  $\text{Nd}_2\text{MgRuO}_6$ , but with partial disorder. To elucidate the magnetic behavior, we employed Monte Carlo (MC) simulations, a powerful computational technique that uses stochastic sampling to model thermodynamic properties and phase transitions in complex systems. The MC results, based on a Heisenberg spin Hamiltonian, reveal ferromagnetic ordering driven by  $\text{Sm}^{3+}$  and  $\text{Ru}^{4+}$  interactions. This work provides crucial insights into the structure-property relationships of  $\text{Sm}_2\text{MgRuO}_6$ , highlighting its potential as a room-temperature ferromagnetic material for spintronic applications. The combination of experimental characterization and computational modeling establishes a robust framework for designing next generation functional perovskites.

## 1 Introduction

Double perovskites of the type  $\text{A}_2\text{B}'\text{B}''\text{O}_6$ , with magnetic  $\text{B}'$  and  $\text{B}''$  ions at the octahedral sites, have gained significant interest due to their potential for magnetoresistive and spintronic applications. Notable examples include  $\text{Sr}_2\text{FeMoO}_6$  exhibiting room-temperature colossal magnetoresistance [1], multiferroic  $\text{Ba}_2\text{FeMnCoO}_6$  [2] and  $\text{Dy}_2\text{NiMnO}_6$  [3], and  $\text{Sr}_2\text{CrIrO}_6$ , a ferrimagnetic insulator with a record-high Curie temperature ( $\sim 90 \text{ K}$ ) that can transition to half-metallicity via doping [4,5]. In contrast, systems with magnetic interactions between A- and B-site cations remain less explored. Ruthenium-based double perovskites are particularly intriguing due to the interplay between 4d electron itinerancy and strong spin-orbit coupling, leading to unique magnetic properties compared to their 3d counterparts [6]. For instance,  $\text{Nd}_2\text{MgRuO}_6$  exhibit coupled A- and B-site spin ordering with antiferromagnetic arrangements [7], though their  $\text{Ru}^{5+}$  ions are magnetically diluted by non-magnetic  $\text{Na}^+/\text{Li}^+$  [8–10]. However, studies on  $\text{Ru}^{4+}$ -based analogues (e.g.,  $\text{Ln}_2\text{NiRuO}_6$ ) are scarce, with reports of rare-earth ordering (Er, Ho) but no long-range Nd magnetism [11]. Here, we present the structural and magnetic properties of  $\text{Sm}_2\text{MgRuO}_6$ , a novel  $\text{Ru}^{4+}$  ( $4d^4$ ) double perovskite stabilized by  $\text{Mg}^{2+}$ . The large ionic radius difference ( $\text{Ru}^{4+}: 0.62 \text{ \AA}$ ,  $\text{Mg}^{2+}: 0.72 \text{ \AA}$ ) drives rock-salt ordering of B-site cations, isolating  $\text{Ru}^{4+}$  and enabling weak magnetic interactions.

## 2 Experimental and Simulation details

### 2.1 Experimental method

The  $\text{Sm}_2\text{MgRuO}_6$  polycrystalline compound was synthesized by conventional high-temperature solid-state reaction method under controlled temperature and time conditions. Stoichiometric proportions of high purity (99.99 %) Samarium (III) oxide ( $\text{Sm}_2\text{O}_3$ ), Ruthenium (IV) oxide ( $\text{RuO}_2$ ) and Magnesium (IV) oxide ( $\text{MgO}$ ) were mixed thoroughly and heated at 800 °C for 12h. The pre-heated powders were sintered again at 1250 °C for 36h with intermediate grinding before any characterization processes could be employed. The XRD pattern with  $\text{Cu-K}\alpha$  radiation ( $\lambda=1.5418 \text{ \AA}$ ) at room temperature shows the presence of a moderately strong superlattice reflection at  $2\theta \sim 19^\circ$  indicating the possibility of crystallographic ordering of the Mg and Ru cations at the perovskite B-site. Rietveld refinement using FULLPROF software indicated that  $\text{Sm}_2\text{MgRuO}_6$  crystallizes in the monoclinic space group  $\text{P}2_1/\text{n}$  that allows for a combination of rock-salt like ordering of the two B-sites. The microstructure, elements distribution and chemical compositions of  $\text{Sm}_2\text{MgRuO}_6$  were analyzed on a scanning electron microscope (SEM) with the Energy Dispersive X-Ray Spectroscopy (EDS) options using an accelerating voltage of 20 kV.

### 2.2 Simulation details

We investigated the magnetic behavior using Monte Carlo (MC) methods, where random sampling of spin configurations reveals equilibrium properties and transition temperatures.

The Hamiltonian describing the system is given by:

$$H = -J_{\text{Sm}-\text{Ru}} \sum_{\langle i,j \rangle} S_i^{\text{Sm}} S_j^{\text{Ru}} \quad (1)$$

where  $J_{\text{Sm}-\text{Ru}}$  is the exchange interaction between Sm-Ru,  $S^{\text{Sm}} = \pm \frac{5}{2}, \pm \frac{3}{2}$  and  $\pm \frac{1}{2}$  is the spin moment of  $\text{Sm}^{3+}$ ,

and  $S^{\text{Ru}} = \pm 1$  the spin moment of  $\text{Ru}^{4+}$ .

By replacing them in eq (1), the energies for different configurations AFM and FM given by:

$$E_{\text{AFM}} = 6\left(\frac{5}{2}\right)J_{\text{Sm}-\text{Ru}} \quad (2)$$

$$E_{\text{FM}} = -6\left(\frac{5}{2}\right)J_{\text{Sm}-\text{Ru}} \quad (3)$$

With  $E_{\text{FM}} = -3578.15960891 \text{ Ry}$  and  $E_{\text{AFM}} = -3578.15174973 \text{ Ry}$

The combination of the Eqs. (2) and (3) gives the value of the coupling interaction between Sm and Ru:

$J_{\text{Sm}-\text{Ru}} = \frac{E_{\text{AFM}} - E_{\text{FM}}}{12\left(\frac{5}{2}\right)}$ , the value of the exchange coupling between obtained by the GGA+U method is

$$J_{\text{Sm}-\text{Ru}} = 5.345 \text{ meV}.$$

## 3 Results and Discussion

### 3.1 Structural Characterization

The XRD pattern of  $\text{Sm}_2\text{MgRuO}_6$  obtained with the Rietveld refinement by the FULLPROF software [11] is plotted in **Fig. 1.** A better fitting result with the  $\text{P}2_1/\text{n}$  space group is realized, and therefore the  $\text{Sm}_2\text{MgRuO}_6$  crystallizes with monoclinic DP-type  $\text{P}2_1/\text{n}$  symmetry. The refinement pattern shows phase-purity and crystallographic information of the oxide without any signature of impurity phases. The values of the reliability factors obtained from the refinement are  $R_{\text{wp}} = 6.4635\%$ ,  $R_{\text{p}} = 5.0538\%$  and  $\chi^2 = 2.89$ . The lattice parameters are  $a = 5.66563(9) \text{ \AA}$ ,  $b = 5.66563(2) \text{ \AA}$ ,  $c = 7.71585(6) \text{ \AA}$  and  $V = 236.584(9) \text{ \AA}^3$ .

The tolerance factor  $t$ :

$$t = \frac{r_{Sm} + r_O}{\sqrt{2} \left( \frac{r_{Mg} + r_{Ru}}{2} + r_O \right)} \quad (4)$$

where  $r$  is ionic radius, can be used to evaluate the structural stability of perovskite-type oxides [12]. A phase stability is confirmed with  $t$  ranging from 0.78 to 1.05 [13]. The value of  $t$  for  $Sm_2MgRuO_6$  is 0.81, which also verifies the stability of DP-type crystal structure.

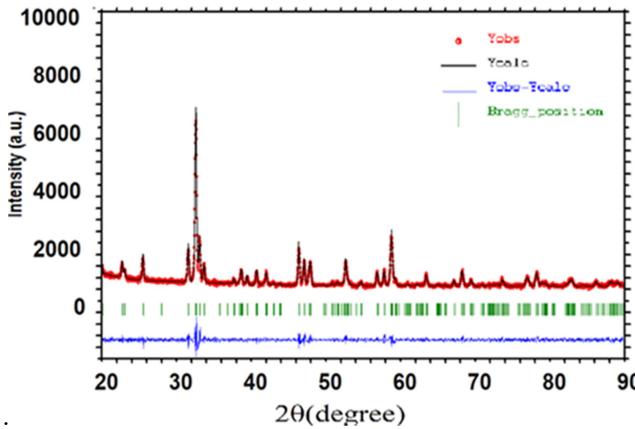


Fig. 1: Powder x-ray diffraction pattern of  $Sm_2MgRuO_6$

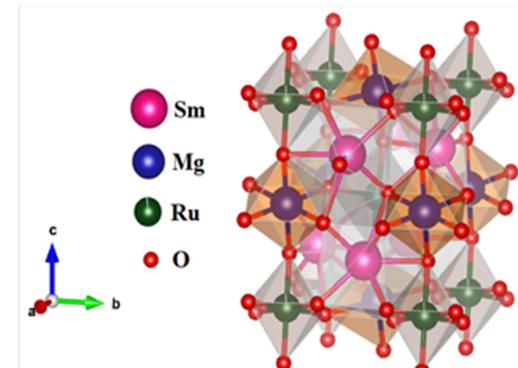


Fig. 2:  $P2_1/n$  Structure

Fig. 2 illustrates the detailed crystal structure of  $Sm_2MgRuO_6$  DP oxide and nearest neighbour environments of each element. In a single ordered unit, it has 20 independent atoms including four Sm, two Mg atoms, two Ru atoms and twelve O atoms. Both Mg and Ru atoms coordinate with six O atoms forming the  $MgO_6$  and  $RuO_6$  octahedrons, respectively. The bond distance ranges and average bond distance are 1.837–2.100 Å and 1.943 Å for  $MgO_6$ . The one for  $RuO_6$  are 1.890–2.174 Å and 2.079 Å for  $MgO_6$  and  $RuO_6$ . The O-Mg-O and O-Ru-O bond angles for  $MgO_6$  and  $TiO_6$  units range from 86.293° to 92.097° and 88.631° to 92.330°, respectively. Sm atom is located in the gap of  $MgO_6/TiO_6$  octahedron forming the 8-coordination polyhedron with the bond distances from 2.279 Å to 2.755 Å. Sm atom occupies the 4e Wyckoff position., Mg and Ru ions are independent positions of 2d and 2c, while there exist three different O ions ( $O_1$ ,  $O_2$  and  $O_3$ ) and they are assumed three different 4e sites.

The atomic positions of Sm, Mg Ru and O are summarized in Table 1 for  $Sm_2MgRuO_6$  DP oxide.

Wyckoff site	Atom	x	y	z
4e	Sm	0.5152(4)	0.5605(5)	0.2462(6)
2d	Mg	1/2	0	0
2c	Ru	0	1/2	0
4e	$O_1$	0.0988(8)	0.4713(3)	1/4
4e	$O_2$	0.3207(7)	0.7281(4)	-0.0477(8)
4e	$O_3$	0.4048(1)	0.9698(0)	0.2641(9)

Table 1. The atomic positions of  $Sm_2MgRuO_6$  DP oxide.

Fig. 3 illustrates the secondary electron SEM image of the typical microstructure together with the element EDS mapping results of  $Sm_2MgRuO_6$  DP oxide. The polycrystalline nature of  $Sm_2MgRuO_6$  is confirmed by the presence of spherical particles with agglomerations in some areas. A number of EDS mapping analysis were performed and fractions confirmed the stoichiometric compositions of 68.0 %, 15.3 %, 12.4 % and 4.4% for Sm, Ru, O and Mg

atoms (Fig.4) respectively. These values are close to the nominal atomic ratio 2:1:1:6 of synthesized  $\text{Sm}_2\text{MgRuO}_6$  oxide.

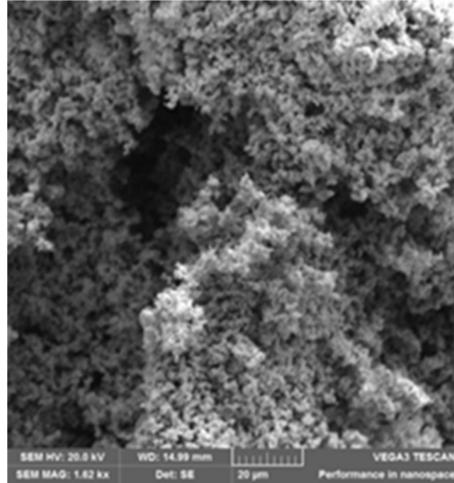


Fig. 3: SEM image

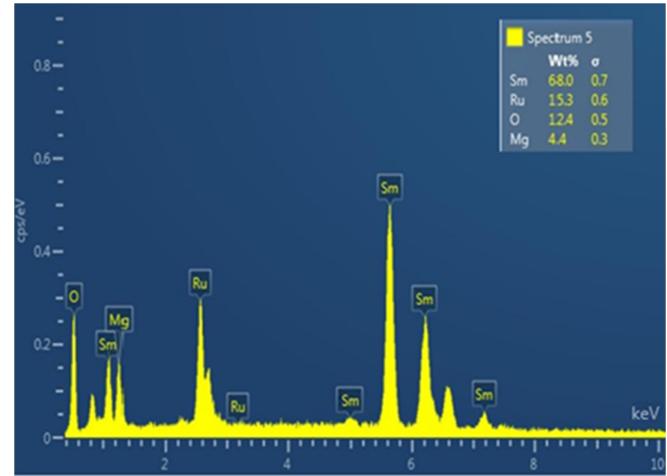


Fig. 4: EDS image

### 3.2 Magnetic Properties: Monte Carlo Simulation

To simulate the magnetic behavior of this complex system a MCS combined with Metropolis algorithm was performed. The system was conceived in the simulation program as a cubic bulk  $L_B = 20$ , which remains lower than the thermodynamic limit determined at  $L_{THL} = 32$  for this class of materials [14], [15]. An initial configuration has been randomly generated in such a way Sm-spins are located randomly in the sites of Sm-sublattice, and Ru-spins are located randomly in the sites of Ru-sublattice. At each MCS step, all the sites in the system are sequentially traversed and single-spin flip attempts are made. A Metropolis algorithm has been included in the program to help deciding the acceptance or rejection of single spin flips. The simulation has recovered the data from 5000 MCS steps, where the first 3000 steps were used to reach the equilibrium. Equations used to calculate the energy; magnetization, magnetic susceptibility ( $\chi$ ) and specific heat ( $C_V$ ) are given respectively by:

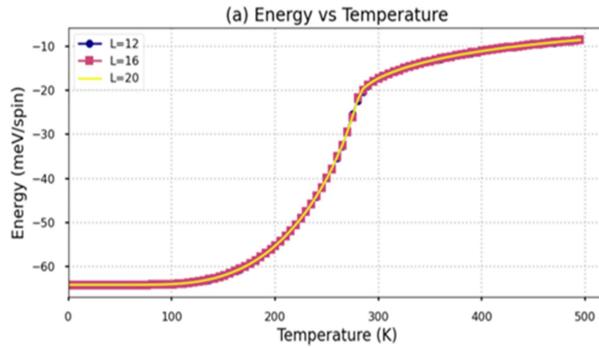
$$E = \frac{\langle H \rangle}{N} \quad (5)$$

$$M = \frac{1}{N} \left| \sum_i S_i^{Sm} - \sum_i S_i^{Ru} \right| \quad (6)$$

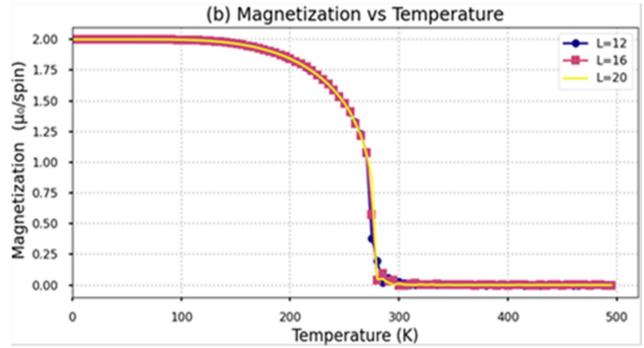
$$C_V = \frac{\beta}{T} (\langle E^2 \rangle - \langle E \rangle^2) \quad (7)$$

with  $\beta = \frac{1}{k_B T}$  where  $T$  is the absolute temperature,  $k_B$  is the Boltzmann constant and  $N$  is the number of spin in the system.

In order to examine the effect of the lattice size, the variation of these quantities with lattice sizes ranging from  $L = 12$  to  $L = 20$  were illustrated in Figs. 5, 6 and 7. Internal energy as a function of temperature is shown in Fig. 5. The simulated transition temperature has been found to be 275 K for small lattice sizes. When the size of the systems increases, the transition temperature reaches a peak at  $T_C = 276.76$  K.

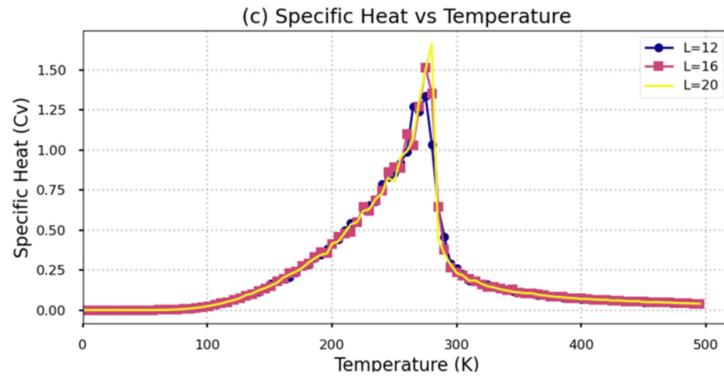


**Fig.5: Internal Energy as a function of temperature for different lattice sizes**



**Fig.6: Total magnetization as a function of temperature for different lattice sizes**

The total magnetization as function of temperature is shown in **Fig. 6** as a function of temperature. The curve reaches a magnetic saturation value of  $M_T = 2\mu\text{B}$ . This magnetization curves tends to gradually decrease after  $T_C$  to reach the minimum value  $M_T = 0,15\mu\text{B}$ . Similar behavior was observed for different lattice sizes. A slight increase in critical temperature of about  $T_C = 276.76$  K were observed. The simulated specific heat as function of temperature ( $C_V$ ) is shown in **Fig.7**. The transition temperature is consistently found at 275 K. For large system sizes, the transition temperature bifurcates to 276.76 K. The curve has a Gaussian form which is a sign of a second order phase transition. Results obtained show that a change in the size of the system is at the origin of a minimal variation in critical temperature  $T_C$ .



**Fig.7: Specific Heat as a function of temperature for different lattice sizes**

To date, the critical temperature  $T_C$  found in this present work has the lowest value compared to the majority of results reported in the literature by other authors in similar studies such as  $\text{La}_{0.67}\text{Sr}_{0.33}\text{MnO}_3$  ( $T_C = 358$  K [16]),  $\text{Mn}_2\text{FeReO}_6$  ( $T_C = 520$  K [17]),  $\text{Sr}_2\text{CrReO}_6$  ( $T_C = 610$  K [18]),  $\text{Sr}_2\text{CrOsO}_6$  ( $T_C = 725$  K [19]),  $\text{Sr}_2\text{CrMoO}_6$  ( $T_C = 458$  K [20]),  $\text{Sr}_2\text{CrWO}_6$  ( $T_C = 500$  K [21]),  $\text{Ca}_2\text{CrMoO}_6$  ( $T_C = 125$  K [22]).

#### 4 Conclusion

In summary, high quality single phased polycrystalline  $\text{Sm}_2\text{MgRuO}_6$  oxide with a B-site ordered monoclinic P21/n (No. 14) DP type crystal structure has been successfully fabricated. A combined theoretical and experimental investigation has been carried out. The  $\text{Sm}_2\text{MgRuO}_6$  oxide belongs to a B-site ordered monoclinic double perovskite (DP) type structure. Monte Carlo simulations confirms the Curie temperature of the  $\text{Sm}_2\text{MgRuO}_6$  to be 276.76 K for

$L=20$ . There are ferromagnetic interactions between  $\text{Sm}^{3+}$  and  $\text{Ru}^{4+}$ , demonstrating how structural distortion governs its unique magnetic properties.

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